# trans-Bis(dimethylphenylphosphine)bis-(3-Z-methoxy-1-propenyl)platinum(II) 

By K. H.P.O'Flynn and W.S. McDonald<br>Department of Inorganic and Structural Chemistry, The University, Leeds LS2 9JT, England

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Abstract. $\left[\mathrm{Pt}\left(\mathrm{PMe}_{2} \mathrm{Ph}\right)_{2}\left(\mathrm{CH}: \mathrm{CHCH}_{2} \mathrm{OMe}\right)_{2}\right]$, $\mathrm{C}_{24} \mathrm{H}_{36} \mathrm{O}_{2} \mathrm{P}_{2} \mathrm{Pt}$, monoclinic, $P 2_{1} / a, a=23 \cdot 26$ (2), $b=$ 6.16 (1), $c=9.28$ (1) $\AA, \beta=107.5$ (2) ${ }^{\circ}, Z=2, D_{x}=1.61$ $\mathrm{g} \mathrm{cm}^{-3}, V=1268 \AA^{3}$. The compound, which was obtained by hydrazine reduction of the corresponding acetylide, is shown to have the $Z$ stereochemistry, with crystallographic symmetry $\overline{1}$.

Introduction. Measurements were made on a Nonius CAD-4 diffractometer using monochromatized $\mathrm{Cu} K \alpha$ radiation. Intensities were measured in the $\theta-2 \theta$ scan mode using a scintillation counter and pulse-height discrimination, and the 1652 independent reflexions which were significantly above background were used in the structure determination. The structure was
solved by the heavy-atom method and refined by fullmatrix least squares. Minimizing $\sum w\left(F_{o}-\left|F_{c}\right|\right)^{2}$, with $w=1 /\left(1+0.002 F_{o}^{2}\right)$, adjustment of coordinates, anisotropic temperature factors for Pt and P , and isotropic temperature factors for C and O , reduced $R$ to $6.89 \%$. Atomic scattering factors, including $\Delta f^{\prime}$ and $4 f^{\prime \prime}$ were taken from International Tables for X-ray Crystallography (1974). The atomic coordinates and vibration parameters with their e.s.d.'s are given in Table 1.*

[^0]Table 1. Fractional coordinates and vibration parameters $\left(\AA^{2} \times 10^{3}\right)$ and their e.s.d.'s The temperature factors are in the form $\exp \left[-2 \pi^{2}\left(U_{11} h^{2} a^{* 2}+U_{22} k^{2} b^{* 2}+U_{33} l^{2} c^{* 2}+2 U_{23} k l b^{*} c^{*}+2 U_{31} l h c^{*} a^{*}+2 U_{12} h k a^{*} b^{*}\right)\right]$

|  | $x$ | $y$ | $z$ | $U_{11}$ | $U_{22}$ | $U_{33}$ | $U_{23}$ | $U_{31}$ | $U_{12}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Pt | 0 | 0 | 0 | 21.2 (4) | $46 \cdot 0$ (4) | $31 \cdot 6$ (4) | $-1 \cdot 1(6)$ | $5 \cdot 7$ (3) | -2.0 (5) |
| P | 0.0840 (2) | -0.1727 (7) | $0 \cdot 1470$ (4) | 28 (2) | 47 (2) | 39 (2) | 1 (2) | 5 (2) | 1 (2) |

Table 1 (cont.)

|  | $x$ | $y$ | $z$ | $U_{\text {iso }}$ |
| :---: | :---: | :---: | :---: | :---: |
| C(1) | $0 \cdot 0122$ (7) | -0.152 (3) | -0.189 (2) | 52 (4) |
| C(2) | $0 \cdot 0398$ (8) | -0.085 (3) | -0.291 (2) | 52 (4) |
| C(3) | $0 \cdot 0668$ (8) | $0 \cdot 134$ (3) | -0.289 (2) | 57 (4) |
| O (4) | $0 \cdot 1261$ (6) | $0 \cdot 104$ (2) | -0.303 (2) | 65 (3) |
| C(5) | $0 \cdot 1541$ (9) | 0.315 (4) | -0.300 (3) | 78 (6) |
| C(6) | 0.0826 (8) | -0.275 (3) | 0.332 (2) | 63 (5) |
| C(7) | $0 \cdot 1075$ (9) | -0.416 (3) | $0 \cdot 066$ (2) | 61 (4) |
| C(8) | $0 \cdot 1505$ (6) | -0.005 (3) | $0 \cdot 191$ (2) | 41 (3) |
| C(9) | $0 \cdot 1779$ (8) | 0.036 (3) | $0 \cdot 080$ (2) | 52 (4) |
| C(10) | 0.2296 (9) | $0 \cdot 169$ (4) | $0 \cdot 109$ (2) | 68 (5) |
| C(11) | $0 \cdot 2525$ (9) | $0 \cdot 267$ (3) | $0 \cdot 251$ (2) | 68 (5) |
| C(12) | $0 \cdot 2251$ (9) | $0 \cdot 230$ (3) | 0.361 (2) | 68 (5) |
| C(13) | $0 \cdot 1729$ (8) | 0.095 (3) | 0.331 (2) | 56 (4) |



Fig. 1. Atom numbering.

Table 2. Bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$ with their e.s.d.'s

| $\mathrm{Pt}-\mathrm{P}$ | $2.282(4)$ | $\mathrm{P}-\mathrm{Pt}-\mathrm{C}(1)$ | $89.2(4)$ |
| :--- | :--- | :--- | :---: |
| $\mathrm{Pt}-\mathrm{C}(1)$ | $2.08(2)$ | $\mathrm{Pt}-\mathrm{C}(1)-\mathrm{C}(2)$ | $132(1)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.35(3)$ | $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $124(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.49(3)$ | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(4)$ | $107(1)$ |
| $\mathrm{C}(3)-\mathrm{O}(4)$ | $1.43(2)$ | $\mathrm{C}(3)-\mathrm{O}(4)-\mathrm{C}(5)$ | $109(1)$ |
| $\mathrm{O}(4)-\mathrm{C}(5)$ | $1.45(3)$ | $\mathrm{Pt}-\mathrm{P}-\mathrm{C}(6)$ | $117.8(6)$ |
| $\mathrm{P}-\mathrm{C}(6)$ | $1.84(2)$ | $\mathrm{Pt}-\mathrm{P}-\mathrm{C}(7)$ | $116.3(6)$ |
| $\mathrm{P}--\mathrm{C}(7)$ | $1.83(2)$ | $\mathrm{Pt}-\mathrm{P}-\mathrm{C}(8)$ | $113.0(5)$ |
| $\mathrm{P}-\square \mathrm{C}(8)$ | $1.80(2)$ |  |  |
|  | $\mathrm{C}-\mathrm{C}$ (benzene ring) $1.38-1.43(3)$ |  |  |

Discussion. The molecular structure and atom numbering are shown in Fig. 1 and bond lengths and angles are given in Table 2. The compound is obtained by the hydrazine reduction of trans-
$\left[\mathrm{Pt}\left(\mathrm{PMe}_{2} \mathrm{Ph}\right)_{2}\left(\mathrm{C} \vdots \mathrm{CCH}_{2} \mathrm{OMe}\right)_{2}\right]$ (Empsall, Shaw \& Stringer, 1975) and this analysis establishes that cis addition occurs, giving the $Z$ stereochemistry shown. The chain of atoms from $\mathrm{C}(3)$ through Pt to $\mathrm{C}\left(3^{\prime}\right)$ is planar, with individual atom displacements of less than one standard deviation. The atoms $\mathrm{Pt}, \mathrm{P}, \mathrm{P}^{\prime}$, $\mathrm{C}(1), \mathrm{C}\left(1^{\prime}\right)$ are necessarily coplanar because of the molecular $\overline{1}$ symmetry, and the dihedral angle between these two planes is $83 \cdot 1^{\circ}$. The differing sizes of Pt and methylene are reflected in the two angles at the olefin
group; the $\mathrm{Pt}-\mathrm{C}(1)-\mathrm{C}(2)$ angle of 132 (1) ${ }^{\circ}$ gives a $\mathrm{Pt} \cdots \mathrm{C}(2)$ contact of $3 \cdot 15 \AA$, while the $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ angle of $124(2)^{\circ}$ gives a $C(1) \cdots C(3)$ distance of $2 \cdot 50 \AA$. These angles may also be influenced by the interaction between Pt and $\mathrm{C}(3)$, which are separated by $3.58 \AA$, with a calculated $\mathrm{Pt} \cdots \mathrm{H}-\mathrm{C}(3)$ contact of $2 \cdot 99 \AA$.

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## References

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International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press.

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# 1-Oxo-2-oxa-3,3-dimethylpent-4-ene-1,4-diylbis(triphenylphosphine)platinum(II) 

By M. C. Norton and W.S. McDonald<br>Department of Inorganic and Structural Chemistry, The University, Leeds LS2 9JT, England

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Abstract. $\left[\mathrm{Pt}\left(\mathrm{CO} . \mathrm{OCMe}_{2} \mathrm{C}: \mathrm{CH}_{2}\right)\left(\mathrm{PPh}_{3}\right)_{2}\right]$, $\mathrm{C}_{42} \mathrm{H}_{38} \mathrm{O}_{2} \mathrm{P}_{2} \mathrm{Pt}$, monoclinic, $P 2_{1} / c, a=12 \cdot 298$ (2), $b=$ 11.038 (3), $c=27.207$ (3) $\AA, \beta=102.66(1)^{\circ}, D_{m}=1 \cdot 53$, $D_{x}=1.533 \mathrm{~g} \mathrm{~cm}^{-3}, Z=4, V=3603 \AA^{3}$. The compound, obtained by carbonylation of a hydroxyacetylene complex, is shown to have a 'platinalactone' structure with a five-membered chelate ring and exocyclic oxo and methylene groups adjacent to the metal.

Introduction. Measurements were made on a Nonius CAD-4 diffractometer using monochromatized $\mathrm{Cu} K \alpha_{1}$ radiation ( $\lambda=1 \cdot 54051 \AA$ ). The cell dimensions and their e.s.d.'s were obtained by a least-squares fit of


Fig. 1. Atom numbering.
$\sin \theta$ values for 25 reflexions centred using the program SETANG. Intensities were recorded in the $\theta-2 \theta$ scan mode using a scintillation counter and pulseheight discrimination. A control reflexion, monitored every 50 reflexions, had fallen in intensity by $16 \%$ at the end of data collection, and the measured reflexions were scaled accordingly. The structure determination used the 3732 independent reflexions with $\theta<70^{\circ}$ and $I>3 \sigma(I)$, where $I=P-2\left(B_{1}+B_{2}\right)$ and $\sigma^{2}(I)=P+$ $4\left(B_{1}+B_{2}\right)+(0.06 I)^{2}$.

The structure was solved by the heavy-atom method and refined by full-matrix least squares using the X-RAY programs (Stewart, Kruger, Ammon, Dickinson \& Hall, 1972). Atomic coordinates, anisotropic temperature factors for Pt and P , and isotropic temperature factors for O and C were refined; the phenyl H atoms were included in calculated positions, assigned the temperature factors of the C atoms to which they are attached, but were not refined. Atomic scattering factors, including $\Delta f^{\prime}$ and $\Delta f^{\prime \prime}$, were taken from International Tables for X-ray Crystallography (1974). Minimization of $w\left(F_{o}-\mid F_{c}\right)^{2}$, with weights derived from the expression for $\sigma^{2}(I)$ given above, gave a final $R$ of $6 \cdot 21 \%$. The atomic coordinates and vibration parameters with their e.s.d.'s are given in Table 1.*

[^1]Table 1. Fractional coordinates ( $\times 10^{4}$ ) and vibration parameters $\left(\AA^{2} \times 10^{3}\right)$ and their e.s.d.'s
The temperature factors are in the form $\exp \left[-2 \pi^{2}\left(U_{11} h^{2} a^{* 2}+U_{22} k^{2} b^{* 2}+U_{33} l^{2} c^{* 2}+2 U_{23} k l b^{*} c^{*}+2 U_{31} l h c^{*} a^{*}+2 U_{12} h k a^{*} b^{*}\right)\right]$ or $\exp \left[-2 \pi^{2} U_{\text {soc }}(2 \sin \theta / \lambda)^{2}\right]$.

|  | $x$ | $y$ | $z$ | $U_{11}$ | $U_{22}$ | $U_{33}$ | $U_{23}$ | $U_{31}$ | $U_{12}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Pt | $1286 \cdot 4$ (5) | $3164 \cdot 6$ (5) | $3553 \cdot 1$ (2) | 41.6 (2) | $36 \cdot 2$ (3) | $41 \cdot 6$ (3) | $8 \cdot 0$ (4) | $17 \cdot 3$ (2) | $10 \cdot 9$ (3) |
| $\mathrm{P}(1)$ | 2718 (3) | 3742 (3) | 3163 (1) | 45 (2) | 38 (2) | 41 (2) | 3 (2) | 18 (2) | 6 (2) |
| P(2) | 2343 (3) | 2504 (3) | 4330 (1) | 44 (2) | 40 (2) | 46 (2) | 7 (2) | 19 (2) | 10 (2) |


[^0]:    * A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 31678 ( 11 pp., 1 microfiche). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

[^1]:    * A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 31679 ( 25 pp., 1 microfiche). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

